



# A Comparative Study of the Physio-Mechanical Properties of Iron Fillings and Mild Steel Chips in Reinforced Particleboard

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## Authors' contributions

This work was carried out in collaboration among all authors. All authors read and approved the final manuscript.

## Article Information

DOI: <https://doi.org/10.9734/jerr/2024/v26i111322>

## Open Peer Review History:

This journal follows the Advanced Open Peer Review policy. Identity of the Reviewers, Editor(s) and additional Reviewers, peer review comments, different versions of the manuscript, comments of the editors, etc are available here:

<https://www.sdiarticle5.com/review-history/126298>

Original Research Article

Received: 25/08/2024

Accepted: 30/10/2024

Published: 05/11/2024

## ABSTRACT

This study compared the physio-mechanical characteristics of mild steel chips and iron fillings on sawdust-produced particleboard to regular particleboard made from sawdust alone, using identical production conditions. Particle board was made using 1.18mm sawdust, 3mm mild steel chips, and iron filler with diameters of 0.15mm, 0.425mm, 0.6mm, 1.18mm, and 2.0mm. 70g of sawdust, 40g of

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**Cite as:** Alozie, Uchenna, Bethel Mba, Allen M. A, Clifford Omonini, S. C. Nwoziri, Franklin Onwuka, Sani Habeeb Muhammed, and Amalime Chukwudi Samson. 2024. "A Comparative Study of the Physio-Mechanical Properties of Iron Fillings and Mild Steel Chips in Reinforced Particleboard". *Journal of Engineering Research and Reports* 26 (11):163-78. <https://doi.org/10.9734/jerr/2024/v26i111322>.

iron fillings, and 40g of mild steel chips were used in the production process. 50ml urea formaldehyde was used as a binder. The atomic absorption spectroscopy was determined for iron fillings and mild steel chips. Particleboards were produced at a temperature of 160°C and a pressure of 20 tons for 15 minutes. Mechanical property tested indicates that particleboard containing iron filings has a lower Modulus of rupture (MOR) because the size of the iron filings in the particleboard reduces as the MOR increases. The particle size of iron filling with the value of 2.0mm had the least MOR of 3.99mpa/m<sup>2</sup> in iron filling samples. The MOR of Mild steel chips was higher compared to all the samples analyzed. Modulus of Elasticity (MOE) of particleboard produced from iron filings increases as the particle sizes of iron filings increases. The samples containing iron filings alone showed the highest MOE of 244.89 MPa/m<sup>2</sup> for 2.0 mm particle board, while the ones containing mild steel chips had the highest MOE of 282.82 MPa/m<sup>2</sup> across all samples, suggesting their strength as reinforcement. The rate of water absorption and thickness of swell of particleboard produced from iron filings increases as the size of the iron filings increase.

**Keywords:** Modulus of rupture; physio-mechanical properties; urea formaldehyde; sawdust; mild steel chips; iron filings.

## ABBREVIATIONS

AAS : Atomic Absorption Spectroscopy  
MOR : Modulus of Rupture  
IF : Iron Filings  
MSc : Mild Steel Chips  
MOE : Modulus of Elasticity  
Sd : Sawdust

## 1. INTRODUCTION

Waste utilization is a desirable substitute since it achieves resource conservation while lowering or even eliminating disposal costs and any pollution issues (Boloy et al. 2021, Kesari et al. 2021, Yuan 2024). Energy and environmental issues must be incorporated with the utilization plan in order to optimize the usage of available resources (Chen et al. 2024). Preserving natural aggregate is crucial to ensuring that future generations will have access to sufficient resources (Li et al. 2024, Reis et al. 2021). Reusing solid waste to partially replace aggregate reduces the need to harvest natural raw materials, which in turn saves landfill space (Colangelo et al. 2020, Dias et al. 2022, Zhang et al. 2024). Researchers and industries are looking for alternative raw materials for composite manufacture, like wood residues and agricultural waste, to replace traditional wood due to the depletion of natural resources and rising demand for wood and wood-based components (Adhikari and Ozarska 2018, Garcia et al. 2024, Pandey 2022, Pedzik et al. 2022). The wood-based production industry now faces new challenges in optimizing the use of available wood and other lignocellulosic raw materials, recycling and reusing wood and wood-based composites (Antov et al. 2023, Kristak et al. 2023, Reh et al. 2022, Yang et al. 2024). There is a great need

for alternative resources because of the expanding environmental concerns and new laws that encourage the cascading use of natural resources (Rehberger and Hiete 2020, Zhou et al. 2023). The scarcity of wood on the local market is a challenge for numerous companies and manufacturers of wood and wood-derived products, leading to intense competition among these industries (Nicholls and Bumgardner 2018). The increasing production capacity that results in more supply in response to the rising demand will make this competition more and more severe. The price of timber could be influenced by a variety of arbitrary circumstances at any given time, which would raise the timber market's cost (Neykov et al. 2020). The raw material handling practices used by the wood-based panel industry are rather flexible because of regional variations in the availability of wood or the constantly fluctuating condition of wood raw materials (Odpes et al. 2021, Sandberg et al. 2023, Sheppard et al. 2020). Furthermore, there has been a notable global growth in demand for wood, which was previously only utilized to manufacture wood-based panels, from other wood-based businesses and the energy industry (Blanchet et al. 2023, Jonsson and Rinaldi 2017, Mantau et al. 2019, Myllyviita et al. 2021). The aforementioned constraints have compelled the wood-based industry to adopt alternative raw resources, such as reclaimed wood and other forest-based material, and to optimize technological manufacturing processes to ensure a stable level of quality (Kutnar 2016, Mariani and Malucelli 2022, Teischinger et al. 2023).

Particleboards are among the most important value-added panel products for the wood-based sector because of their wide range of applications (Claisse 2016, Vale 2017).

Particleboard manufacturing can use lower-quality raw materials than the pulp and paper sector or building (Marasilp and Sangkakool 2024, Zabowski et al. 2024). For the sake of the environment, proper waste management is crucial, especially with regard to wood and wood-based byproducts (Abu Bakar and Salim 2023, Dey et al. 2024, Mohammad 2023). One potential solution to mitigate the anticipated decline in the technical qualities of the boards resulting from the incorporation of different lignocellulosic raw materials is to add more binder (Ferreira et al. 2019a, Ferreira et al. 2019b, Gigar et al. 2023, Melichar et al. 2022). The type and concentration of resin employed are important factors in determining the right properties and intended uses of particleboard binding (Hussin et al. 2022, Solt et al. 2019). If the board must have a high degree of water resistance, replace UF with a different resin, such as PF (phenol-formaldehyde) or PMDI (polymeric 4,4'-methylenediphenyl isocyanate) (Esmaili et al 2017, Sarika et al. 2020). This is standard practice when working with particles and will guarantee optimal bonding and strength parameters. The utilization of substitute raw materials not only reduces the cost of producing panels but also contributes to the sustainable management of leftover forest biomass that isn't being put to better use (Arâmburu et al. 2022, Simões et al. 2023).

The sawdust, wood chips, shavings, pulp, and other wastes from sawmill operations, together with bark, harvested and generated wastes, and unprocessed sawmill by-products, are some examples of replacement raw materials (Asamoah et al. 2020). Wood leftovers and byproducts are being used increasingly extensively in industry to make wood-based panels, which is justified by the growing shortage of wood raw materials like roundwood and full-value wood (Gardiner and Moore 2014, Jakes et al. 2016). Particleboards can be made from one or more replacement raw resources, such as post-consumer wood, wood from fruit trees, and wood from urban greenery, crushed into lignocellulosic particles (Dukarska et al. 2021). One practical way to produce boards fit for furniture and interior uses is to recycle debris from building and demolition into residual medium density fiberboard (MDF), particleboard, cardboard, and plywood (Alsheyab 2022). Particle board primarily composed of wood chips or fragments mixed with a suitable binder (such as synthetic resin) and heated to high pressure in a hot press to fuse together (Orelma et al. 2021,

Pizzi 2016, Pizzi et al. 2020). The entire interparticle bond between the particles is formed by the addition of binder, and other materials may have been added afterwards to improve specific features of the production of particleboard (Scharf et al. 2023). The pressing process gives particleboards additional characteristics (Farrokhpayam et al. 2016, Gonçalves et al. 2020, Laskowska 2024). The qualities of binding that are employed to fuse the materials together during particleboard processing also have an impact on the attributes, in addition to the composition and structure that can be obtained with these elements (Borges et al. 2023). Sawdust is a crucial ingredient used in the production of particle board (Olaiya et al. 2023). The capacity of the sawdust size to allow bonding materials to distribute throughout the boards, enhancing the bonding quality, which in turn enhances the mechanical properties of the produced material. Equal mechanical qualities are maintained throughout the created board due to the size, which permits easy and equal spreading with additional material (El Hamri et al. 2024). Traditional synthetic adhesives used in the production of wood-based panels include urea, formaldehyde, phenol, melamine, and others (Calvez et al. 2024, Ferdosian et al. 2017, Gonçalves et al. 2021, Hemmilä et al. 2017). These adhesives are made from materials obtained from petroleum (Wu et al. 2024). Addition of thermosetting resin treatment to wood (sawdust) and other particles improves the mechanical characteristics of particleboards (Altgen et al. 2015).

## 2. MATERIALS AND METHODS

### 2.1 Materials and Equipment

Sawdust of 1.18mm diameter, Urea formaldehyde resin, Iron fillings of sizes 0.15mm, 0.425mm, 0.6mm, 1.18mm, 2.0mm respectively and Mild steel chip of 3mm thickness, metal mold of dimension 12cm x 10cm x 3cm, a digital multi-thermometer of temperature range (-50 to 300°C), Water used for testing water absorption and thickness of swelling, and urea formaldehyde. Schmidt hammer for testing the compressive strength of the particleboard, Artist saw used for cutting of the particleboard, the universal testing machine (500 tons) for testing the breaking load (KN) of the specimen, Vernier caliper used for obtaining the thickness of the particleboard, digital weighing balance and an electronic multi thermometer. Fig. 1, Fig. 2 and Fig. 3 represents samples of iron fillings, samples of sawdust and Schmidt hammer respectively.



Fig. 1. Iron filings sample



Fig. 2. Sawdust sample



Fig. 3. Schmidt Hammer



Fig. 4. Weighted Sawdust sample



Fig. 5. Weighted Iron filings sample

## 2.2 Method

### 2.2.1 Collection of raw materials

The raw materials used were sawdust, iron filings, mild steel chips and urea formaldehyde. The sawdust used for the investigation was obtained directly from a sawmill factory after which it was dried in an open air to remove the moisture content to about 6% (Salam et al. 2024, Silva et al. 2021). The saw dust was sieved to obtain a particle size of 1.18mm using a sieve of 1.18mm in Civil Engineering Department of Michael Okpara University of Agriculture (Zepeda-Cepeda et al. 2021). The required quantity saw dust weight were weighed in the weighing balance to obtain 800g of sawdust after which it was kept in a cool dry place. The mild steel chips used for the study was obtained from the mechanical engineering workshop after machining the mild steel shaft of 25mm diameter with a feed of 3mm to avoid obtaining a rusted chip. The steel chips were weighed to obtain 100g. the iron filings used were also obtained

from the workshop using a chip gotten after machining a sample of 25mm diameter sample. The iron filings were sieved to obtain a particle size of 0.15mm, 0.425mm, 0.6mm, 1.18mm and 2.0mm respectively using the corresponding sieve size. The iron filling was weighed to obtain 60g for each particle size obtained. Fig. 4 and Fig. 5 represents weighted sawdust sample and weighted iron filings sample.

### 2.2.2 Determination of the atomic absorption spectroscopy (AAS) of iron fillings and mild steel chips

The determination of the atomic absorption spectroscopy was carried out with the aim of determining the chemical elements constituent in the iron fillings and mild steel chip. 10g of the prepared samples of iron fillings and mild steel chips were measured respectively. Each of the samples were put in two different beakers and mixed with about 50ml of concentrated nitric acid ( $\text{HNO}_3$ ) and perchloric acid. The mixture was put in the fume cupboard digester model ISOCIDE

(Frontier Junior) for further digestion for about 24hrs. The digested particles are further taken to the spectrophotometer where it was atomized and the chemical element composition were picked up by the cathode lamps. Fig. 6 represents the atomic absorption spectroscopy setup.

### 2.2.3 Urea formaldehyde

The urea formaldehyde resin that was used as the binding agent was prepared in the department of Chemistry Michael Okpara University of Agriculture. The raw material used in this preparation were 100g of Urea, 200g of Formaldehyde, 80g of Sodium hydroxide, 100ml of Acetic acid 100ml and Distilled water. The urea formaldehyde was prepared by adjusting the pH scale of 200g formaldehyde (38%) to 7.5 with an aqueous solution of sodium hydroxide NaOH (10%) and 100g urea was added. The mixture was stirred and heated with reflux for 1½ hours. Then about 30ml of water was distilled off to obtain a resin of 60% solids and 2.5% free formaldehyde. Thus, obtain liquid resin was treated with 0.3N acetic acid to Ph of 4.0 and cured under reflux for the purpose of proper dissolution of urea particles, this was done for additional 2hours. A gelled product was found and was dried at about 45°C for 1hr to remove the moisture content after which it was kept in at room temperature.

### 2.2.4 Production of particleboard

Seven distinct containers were filled with precisely measured 70g of the processed sawdust, in preparation for the blending operation. A laboratory beaker was used to measure 50ml of urea formaldehyde into seven distinct plastic containers. 40g of each of the iron fillings with sizes (0.15mm, 0.425mm, 0.6mm,

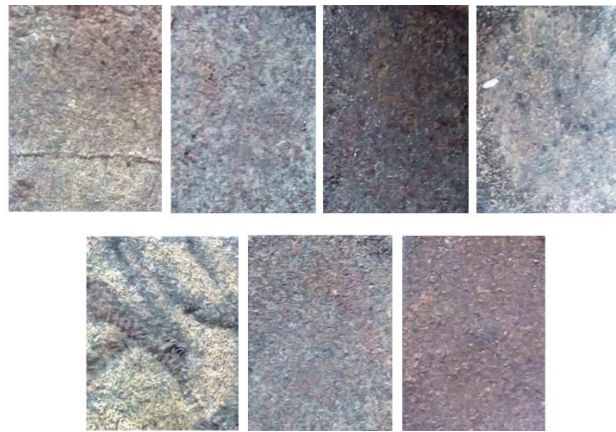
1.18mm and 2.0mm) respectively were measured and kept ready for blending. 40g of the mild steel chips were also measured and kept ready for blending. Prior to the production process, 50ml of urea formaldehyde and 70g of sawdust were carefully combined to improve uniformity. The mixture was spread out on a metal mold measuring 12 cm by 10 cm by 3 cm. The mold was covered with aluminum foil to make removal easier and prevent the particleboard from burning, which was a crucial feature due to the foil's high heat resistance. Sawdust was placed between the metal mold and covered with folded aluminum foil. A mechanical stirrer was used to completely mix 70g of the prepared sawdust and 50ml of urea formaldehyde with each of the determined iron filler particle sizes. The 40g mild steel chip was also combined with 50ml of urea formaldehyde resin and used to reinforce one of the 70g sawdust. For every sample, the mixing of the iron filler and 70g of sawdust was repeated. Every generated sample was dispersed throughout the atmosphere for a further three hours of drying. Ultimately, the resulting particleboard was stored for testing. Fig. 7 shows the produced samples of particleboard.

### 2.2.5 Produced particleboard testing methods

**Compressive strength:** The compressive strength of the particleboards produced above was determined using the Schmidt hammer. The manufactured boards were cut into a rectangular shape of dimensions 50mm x 60mm and the thickness of each sample were measured and recorded. The prepared samples were fixed on a grip and the Schmidt hammer was used to indent on the surface of the board and the rebound number was noted and recorded, the corresponding compressive strength was read off from the rebound chart.



Fig. 6. Atomic absorption spectroscopy (AAS) setup



**Fig. 7. Produced particleboard**

**Tensile strength:** The tensile strength of the particleboards produced above was determined using the universal testing machine (UTM). A sample with a 50 x 60 mm rectangular form was placed in the grip of a universal testing machine with a 500-ton capacity, and it was then automatically loaded from the computer. The graph of load/ extension was plotted on the computer interface after which it was extracted for evaluation.

**Modulus of rupture (MOR):** The modulus of rupture is a crucial mechanical property of particleboard that influences its ability to burst when a load is applied. This was derived using each sample's failure load value that was found. as a consequence, applying the formula;

$$MOR = \frac{3PL}{2bd^2}$$

Where;

$P$  = Failure load (N)

$L$

= The board span between the machine support

$b$  = Width of the board sample (mm)

$d$  = Thickness of the board sample (mm)

**Water absorption test:** A top-loading digital weighing balance was used to weigh a sample of each particleboard that was manufactured. The samples were immersed in water for 24 hours before being weighed. The percentage that indicates the water absorption is the weight difference relative to the sample's initial weight. For every newly established board, this procedure was carried out.

$$Water\ absorption = \frac{W_2 - W_1}{W_1} \times \frac{100}{1}$$

Where;

$W_1$  = Weight of sample before soaking

$W_2$  = Weight of sample after soaking

**Thickness swelling test:** The Veneer Caliper was used to measure the thickness of each created board both before and after it was soaked in water for 24 hours. The thickness swell, which was calculated for each of the above-formed boards, is the ratio of the variations in thickness to the sample's initial thickness represented as a percentage.

$$Thickness\ swell = \frac{T_2 - T_1}{T_1} \times \frac{100}{1}$$

Where;

$T_1$  = Thickness of sample before soaking

$T_2$  = Thickness of sample after soaking

**Density test:** The density of a particleboard is one the important parameters to be determined and the targeted value is between 0.6 to 0.8g/cm<sup>3</sup>. The density was determined by dividing the mass of the board sample before soaking, by the volume of the sample. This complies with ANSI A208.1-1999, the American National Standard Institute's code. The aim is to produce standard board particles with densities ranging from 37lb/ft<sup>3</sup> to 50lb/ft<sup>3</sup> which is equivalent to 0.6g/cm<sup>3</sup> to 0.8g/cm<sup>3</sup>.

$$Density = \frac{mass}{volume}$$

Where;  
 Volume of sample ( $cm^3$ )  
 = length  $\times$  width  $\times$  thickness

Mass (g)  
 = mass of sample before soaking in water

### 3. RESULTS AND DISCUSSION

#### 3.1 Results

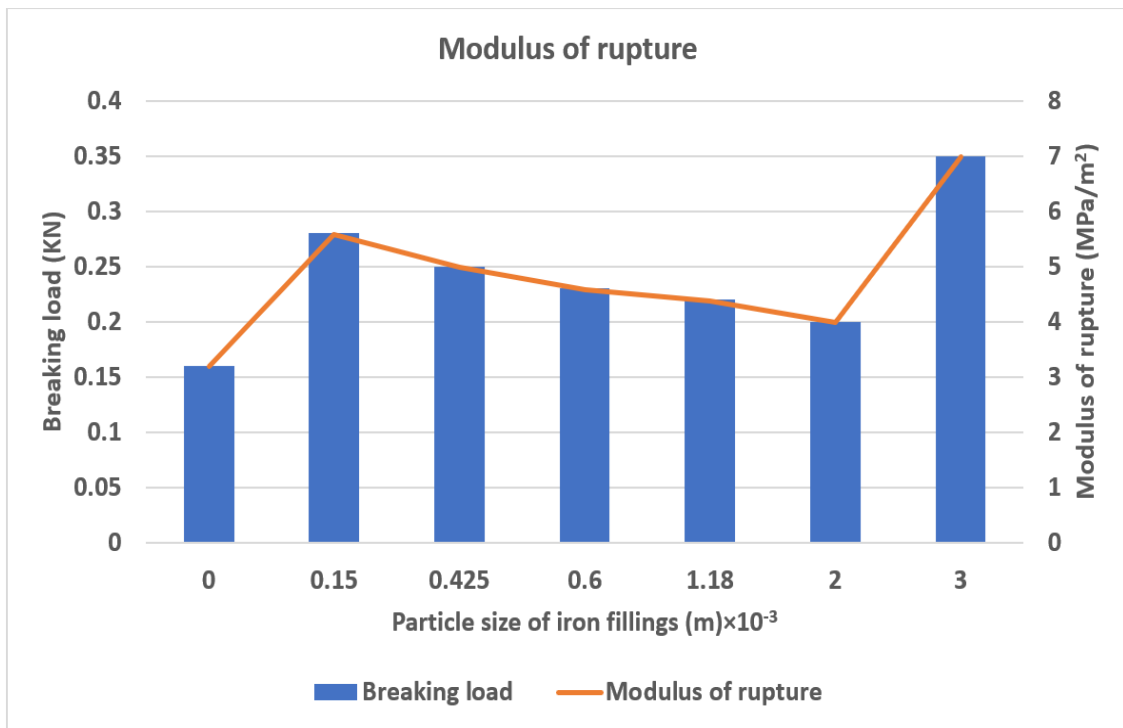


Fig. 8. Modulus of rupture according to applied load

Table 1. Chemical compositions of Iron filings and Mild steel Chips (%)

Elements	Iron filings	Mild steel chip
Ca	2.001	6.802
CaO	3.80	9.516
Mg	0.973	1.763
MgO	1.613	2.923
LOi	1.48	6.34
K	1.25	1.425
K <sub>2</sub> O	1.506	1.717
Na	5.525	5.425
NaO	7.45	7.313
Fe	11.08	12.07
Fe <sub>2</sub> O <sub>3</sub>	15.73	17.14
Zn	11.21	15.54
Zn <sub>2</sub> O	13.95	22.46
MC	0.96	2.34
Pb	4.71	4.58
Mn	6.08	6.02

**Table 2. The dimensions of the tested sample**

Mat	Particle size of iron fillings (m) $\times 10^{-3}$	Length of the sample (m)	Width of the sample (m)	Thickness of the sample (m)	Cross-sectional area (m)
<b>Sd</b>	0 [Control]	0.06	0.05	0.0095	$4.75 \times 10^{-4}$
<b>IF</b>	0.15	0.06	0.05	0.0095	$4.75 \times 10^{-4}$
	0.425	0.06	0.05	0.0095	$4.75 \times 10^{-4}$
	0.6	0.06	0.05	0.00951	$4.75 \times 10^{-4}$
	1.18	0.06	0.05	0.00952	$4.75 \times 10^{-4}$
	2.0	0.06	0.05	0.009523	$4.75 \times 10^{-4}$
<b>MSc</b>	3.0	0.06	0.05	0.009525	$4.75 \times 10^{-4}$

**Table 3. The modulus of Rupture results for the tested samples**

Mat	Particle size of iron fillings (m) $\times 10^{-3}$	Length of the sample [L] (m)	Width of the sample [b] (m)	Depth [d] (m)	Breaking load [P] (KN)	Modulus of rupture (MPa/m <sup>2</sup> )
<b>Sd</b>	0 [control]	0.06	0.05	0.0095	0.16	3.19
<b>IF</b>	0.15	0.06	0.05	0.0095	0.28	5.58
	0.425	0.06	0.05	0.0095	0.25	4.98
	0.6	0.06	0.05	0.00951	0.23	4.58
	1.18	0.06	0.05	0.00952	0.22	4.38
	2.0	0.06	0.05	0.009523	0.20	3.99
<b>MSc</b>	3.0	0.06	0.05	0.009525	0.35	6.98

**Table 4. The density of the tested sample**

Mat	Particle size of iron fillings (m) $\times 10^{-3}$	Weight (g)	Length (cm)	Width (cm)	Thickness (cm)	Volume (cm <sup>3</sup> )	Density (g/cm <sup>3</sup> )
<b>Sd</b>	0 [Control]	20.53	6	5	0.95	28.5	0.72
<b>IF</b>	0.15	22.21	6	5	0.95	28.5	0.77
	0.425	21.571	6	5	0.95	28.5	0.75
	0.6	23.38	6	5	0.951	28.53	0.81
	1.18	24.219	6	5	0.952	28.56	0.84
	2.0	24.20	6	5	0.952	28.56	0.84
<b>MSc</b>	3.0	24.40	6	5	0.952	28.56	0.74

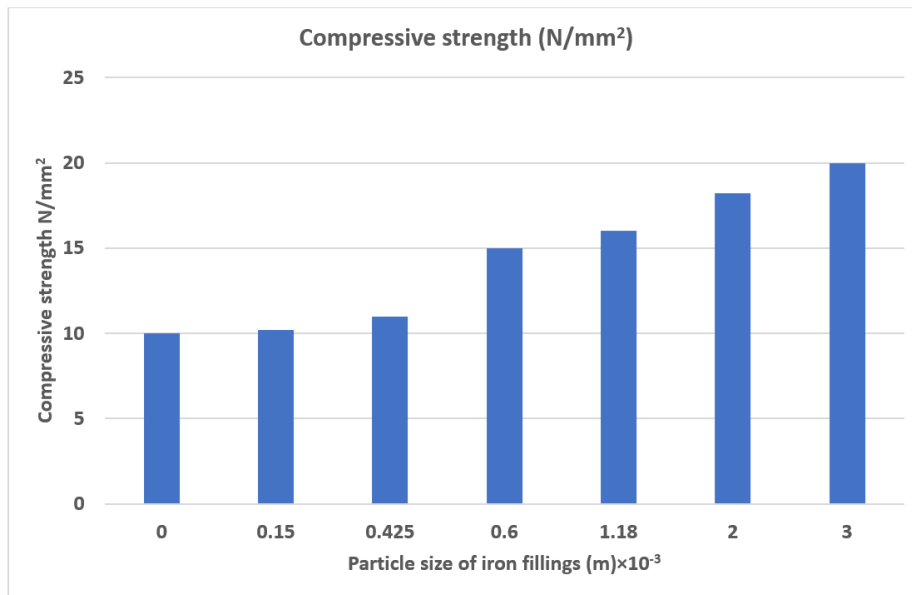
**Table 5. The compressive strength of sizes of iron filings and Mild steel Chip**

Particle size of iron fillings (m) $\times 10^{-3}$	Compressive strength N/mm <sup>2</sup>
0 [Control]	10.00
0.15	10.200
0.425	11.00
0.6	15.00
1.18	16.00
2.0	18.200
3.0 [Mild steel]	20.00



**Table 6. Water absorption of the iron filings and Mild steel Chips particleboards**

Mat	Particle size of iron fillings (m)×10 <sup>-3</sup>	Weight before soaking in water (g)	Weight after soaking in water (g)	Water absorption %
<b>Sd</b>	0 [control]	20.53	32.13	56.5
<b>IF</b>	0.15	22.21	33.10	49
	0.425	21.57	32.2	49.2
	0.6	23.38	34.92	49.4
	1.18	24.21	37.12	53.3
	2.0	24.20	37.45	54.75
<b>MSc</b>	3.0 [Mild steel chip]	21.40	33.25	55.57



**Fig. 9. Compressive strength according to particle size**

**Table 7. Thickness of swell of iron filings and mild steel chip particle boards**

Mat	Particle size of iron fillings (m)×10 <sup>-3</sup>	Thickness before soaking in water (mm)	Thickness after soaking in water (mm)	% thickness of swell
<b>Sd</b>	0 [Control]	9.500	13.525	42.36
<b>IF</b>	0.15	9.500	10.600	11.57
	0.425	9.500	10.80	13.68
	0.6	9.510	11.00	15.66
	1.18	9.520	10.5	10.29
	2.0	9.523	12.10	27.06
<b>MSc</b>	3.0	9.525	13.50	41.7

**Table 8. Modulus of Elasticity of the tested samples**

Mat	Particle size of iron fillings (m)×10 <sup>-3</sup>	Tensile strength (MPa)	Strain (e)	MOE (MPa/m <sup>2</sup> )	% Elongation
<b>Sd</b>	0 [Control]	2	0.0266	75.18	2.66
<b>IF</b>	0.15	4	0.045	88.88	4.5
	0.425	7	0.05	140	5
	0.6	16	0.076	210.52	7.6
	1.18	20	0.0866	230.95	8.6
	2.0	24	0.098	244.89	9.8
<b>MSc</b>	3.0	28	0.099	282.82	9.9

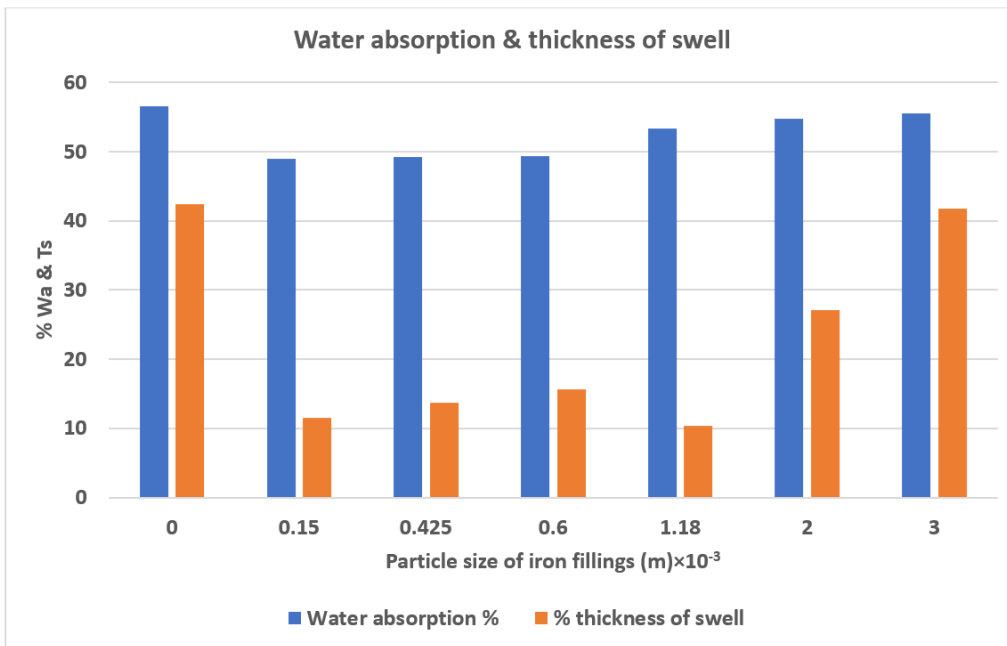


Fig. 10. Water absorption & thickness of swell according to particle size

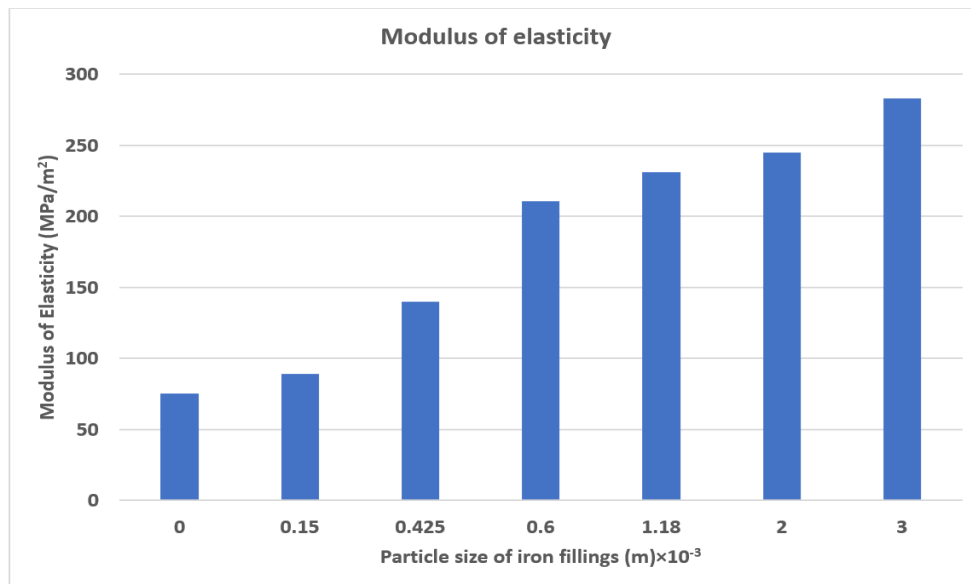


Fig. 11. Modulus of elasticity according to particle size

### 3.2 Discussion

Table 1 shows the result of the atomic absorption spectroscopy obtained for iron filings and mild steel chips. It can be observed that the iron filings contain 11.08% Fe, 11.21% Zn, 6.08% Mn, 1.25%K, 4.71% Pb, 5.525% Na, 2.01% Ca, 0.973% Mg and their corresponding oxides. While the mild steel chips contain 6.812%Ca, 1.763%Mg, 1.425%K, 5.425%Na, 12.07%Fe, 15.54%Zn, 4.58%Pb, 6.02%Mn and their

corresponding oxides. Table 2 shows that the thickness of samples containing mild steel chip and iron filing with sizes of 0.6 mm, 1.18 mm, and 2.0 mm increased to 9.51 mm, 9.52 mm, 9.523 mm, and 9.525 mm. However, because of the huge filler particle size and the hydraulic press's strong compressive strength, the results were the same for the control and samples with 0.15mm and 0.425mm iron filings. Table 3 shows the modulus of rupture of the samples and it was observed that the MOR of sample with mild steel

chip was highest with 6.98MPa/m<sup>2</sup> MOR. The least is the control sample with 3.19MPa/m<sup>2</sup> whereas in samples with iron filings the MOR decreases with respect to increase in particle sizes, having values of 5.58MPa/m<sup>2</sup>, 4.98MPa/m<sup>2</sup>, 4.58MPa/m<sup>2</sup>, 4.38MPa/m<sup>2</sup> and 3.99MPa/m<sup>2</sup> for iron filing particle sizes of 0.15mm, 0.425mm, 0.6mm, 1.18mm and 2.0mm respectively.

It was noted that sample with mild steel chip had the highest compressive strength of 20MPa/mm<sup>2</sup> while that of samples with iron filings increase gradually as the particle size increases as shown in Fig. 9. For the water absorption, particleboard without iron filings (control) maintained the highest water absorption rate of 56.5% as seen in Table 6 while particleboard with 0.15mm, 0.425mm, 0.6mm, 1.18mm, and 2.0mm iron filings has water absorptions of 49%, 49.2%, 49.4%, 53.3%, 54.75%, and 55.37% respectively. Based on the findings, we can infer that the rate of water absorption owing to porosity increases with the size of iron filled particles. Conversely, mild steel chip particleboard likewise has a high-water absorption rate of 55.37%. Fig. 10 illustrates the samples' swell thickness. As compared to the samples with iron filing sizes of 0.15mm, 0.425mm, 0.6mm, 1.18mm, and 2.0mm, respectively, the control sample exhibited a high value thickness of swell (42.36%). The samples also demonstrated a progressive increase in swell. Additionally, it is observed that the iron filing sample exhibited a sudden decrease in thickness followed by an increase; this is due to the iron filing and sawdust particles having similar particle sizes. However, compared to mild steel chip, particleboard containing iron filings has a better swell thickness.

Lastly, according to 11, the sample containing mild steel chips from Table 8 has the highest modulus of elasticity, measuring 282.82 MPa/m<sup>2</sup>, while the sample containing iron filings shows a gradual increase, going from 88.88 MPa/m<sup>2</sup> to 140 MPa/m<sup>2</sup>, 210.52 MPa/m<sup>2</sup>, 230.95 MPa/m<sup>2</sup>, and 244.89 MPa/m<sup>2</sup> at the sizes of 0.15mm, 0.425mm, 0.6mm, 1.18mm, and 2.0mm, respectively. However, the control has the lowest modulus of elasticity of 75.18 MPa/m<sup>2</sup>.

#### 4. CONCLUSION

This paper Examines the comparative study of the Physio- Mechanical Properties of Iron filling and Mild Steel Chips in Reinforce Particle board.

The following conclusions were drawn based on the results obtained from this research work.

1. Mechanical property tested indicates that particleboard containing iron filings has a lower modulus of rupture (MOR) because the size of the iron filings in the particleboard reduces as the MOR increases. The particle size of iron filling with the value of 2.0mm had the least MOR of 3.99mpa/m<sup>2</sup> in samples of iron filling alone. The MOR of Mild steel chips was higher compared to all the samples analyzed.
2. Modulus of elasticity (MOE) of particleboard produced from iron filings increases as the particle sizes of iron filings increases. Particle 2.0mm has the highest MOE 244.89mpa/m<sup>2</sup> in samples of iron filings only while samples of Mild steel chips were seen to have the highest value of 282.82mpa/m<sup>2</sup> of MOE compared to all the samples indicating its strength as reinforcement.
3. The rate of water absorption and thickness of swell of particleboard produced from iron filings increases as the size of the iron fillings increase. The water absorption rate was high with mild steel chips due to porosity as a result of weak inter-particle bond between the chips and the sawdust particle.
4. The significance of this research is to obtain particle board with good strength, smooth surface and better resistance to swelling. It is advised to utilize a homogenous material that is highly slender (long, thin particles), free of dust, splinters, and oversize particles.

#### DISCLAIMER (ARTIFICIAL INTELLIGENCE)

Author(s) hereby declares that NO generative AI technologies such as Large Language Models (ChatGPT, COPILOT, etc.) and text-to-image generators have been used during writing or editing of this manuscript.

#### ACKNOWLEDGEMENT

The authors would like to extend their sincere gratitude to Michael Okpara University of Agriculture Mechanical Engineering Department for their invaluable assistance in granting them access to the laboratory's supplies and equipment.

## COMPETING INTERESTS

Authors have declared that no competing interests exist.

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